

# THEORETICAL MODEL IN CYCLIC VOLTAMMETRY OF AN ELECTRODE REACTION OF WATER-SOLUBLE REDOX ENZYMES ASSOCIATED WITH REVERSIBLE REGENERATIVE STEP- MATHCAD SIMULATION FILE for Studying kinetics and thermodynamics of Enzyme-Substrate and Drug-Drug interactions

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## **Abstract**

In our recent work published in *Croatica Chemica Acta* 92 (4) (2019) 1-8 we reported on a new theoretical model of a diffusional EC' mechanism associated with reversible regenerative step. Model is suitable for assessing the kinetic and thermodynamic parameters relevant to hydrophilic enzymes (as cytochromes) and hydrophilic substrates. The features of simulated voltammetric patterns are function of parameters related to the electrode reaction (standard rate constant of electron transfer, number of exchanged electrons, electron transfer coefficient), and they additionally depend on the kinetic and thermodynamic of regenerative reaction. We provide the readers entire simulation protocol in MATHCAD for calculation of theoretical cyclic voltammograms. Model is also relevant to study many hydrophilic drug-drug interactions. In the work published in *Analytical and Bioanalytical Electrochemistry* 12 (2020) 345-364, we provide a large set of experimental results relevant to drug-drug and drug-DNA interactions evaluated from the models we present in that work.

$$\tau_{ac} := 0.01 \quad \tau := 0.01 \quad D := 0.000005$$

$$el := 2 \quad \alpha := 0.5 \quad k_s := 0.02 \quad k_c := 0.01 \quad d := \frac{\tau}{25}$$

$$E_s := -0.4 \quad E_f := 0.4 \quad \Delta E := E_f - E_s \quad dE := 0.004 \quad s := 1 \cdot \frac{\tau_{ac}}{d}$$

$$m := \frac{\tau}{d} + 1 \cdot \frac{\Delta E}{dE} \cdot 25 + \frac{\tau}{d} \quad n := \frac{\Delta E}{dE} \cdot 25 + \frac{\tau}{d} + 1 \cdot \left( \frac{\Delta E}{dE} \cdot 25 \cdot 2 + \frac{\tau}{d} \right)$$

$$E_m := E_s + \left( \text{ceil} \left( \frac{m - \frac{\tau}{d}}{25} \right) \cdot dE - dE \right)$$

$$k := 1 \cdot 2 \cdot \left( \frac{\Delta E}{dE} \cdot 25 + \frac{\tau}{d} \right)$$

$$E_n := E_f - \left[ \text{ceil} \left( \frac{n - \left( \frac{\Delta E}{dE} \cdot 25 + \frac{\tau}{d} \right)}{25} \right) \cdot dE - dE \right]$$

$$R := 8.314 \quad F := 96500$$

$$T := 298$$

$$\Phi_{ac} := \frac{el \cdot F}{R \cdot T} \cdot E_f \quad \Phi_{em} := \frac{el \cdot F}{R \cdot T} \cdot E_m \quad \Phi_n := \frac{el \cdot F}{R \cdot T} \cdot E_n \quad S_k := \text{erf} \left( \frac{\sqrt{K_{catalytic}} \cdot \sqrt{k}}{\sqrt{50}} \right) - \text{erf} \left( \frac{\sqrt{K_{catalytic}} \cdot \sqrt{k-1}}{\sqrt{50}} \right)$$

$$\Psi_s := \frac{KET \cdot e^{-\alpha \cdot \Phi_{ac}}}{1 + KET \cdot e^{-\alpha \cdot \Phi_{ac}} \cdot \left( 1 + e^{\Phi_{ac}} \right) \cdot \frac{S_1 \cdot K_{eq}}{K_{catalytic} \cdot (1 + K_{eq}) \cdot 25} + KET \cdot e^{-\alpha \cdot \Phi_{ac}} \cdot \left( 1 + e^{\Phi_{ac}} \right) \cdot \frac{S_1 \cdot 1}{25 \cdot (1 + K_{eq})}}$$

## EC catalytic mechanism with reversible regenerative reaction

### Symbols and abbreviations

KET-is dimensionless kinetic parameter of electron transfer

$k_s$  -is standard rate constant of electron transfer

$k_s$  - is rate constant of catalytic reaction

$K_{eq}$  - is equilibrium constant of regenerative chemical reaction

$K_{catalytic}$  - is dimensionless catalytic parameter

$D$  - is diffusion coefficient

$\alpha$  - is electron transfer coefficient

$el$  - is number of electrons

$\tau$  - is time frame of potential steps in cyclic staircase voltammetry

$d$  - is the time increment

$dE$  - is potential step height

$E_s$  - is starting potential

$E_f$  - is final potential

$E_m$  and  $E_n$  are potential ramps of cathodic and anodic scan, respectively

$R$ - is gas constant

$T$  - is thermodynamic temperature

$\Phi_m$   $\Phi_{ac}$  and  $\Phi_n$  are dimensionless potentials

$\Psi$  is dimensionless current

$S_k$  is numerical integration factor

$\Delta E$  is potential frame

$\Psi_c$  is cathodic current

$\Psi_a$  is anodic current

$$\Psi_m := \frac{\text{KET} \cdot e^{-\alpha \cdot \Phi_m} \left[ 1 - \frac{(1 + e^{\Phi_m}) \cdot \text{Keq}}{\text{Kcatalytic} \cdot (1 + \text{Keq}) \cdot 25} \cdot \sum_{j=1}^{m-1} (\Psi_j \cdot S_{m-j+1}) - \frac{(1 + e^{\Phi_m}) \cdot 1}{25 \cdot (1 + \text{Keq})} \cdot \sum_{j=1}^{m-1} (\Psi_j \cdot S_{m-j+1}) \right]}{1 + \text{KET} \cdot e^{-\alpha \cdot \Phi_m} \cdot (1 + e^{\Phi_m}) \cdot \frac{S_1 \cdot \text{Keq}}{\text{Kcatalytic} \cdot (1 + \text{Keq}) \cdot 25} + \text{KET} \cdot e^{-\alpha \cdot \Phi_m} \cdot (1 + e^{\Phi_m}) \cdot \frac{S_1 \cdot 1}{25 \cdot (1 + \text{Keq})}}$$

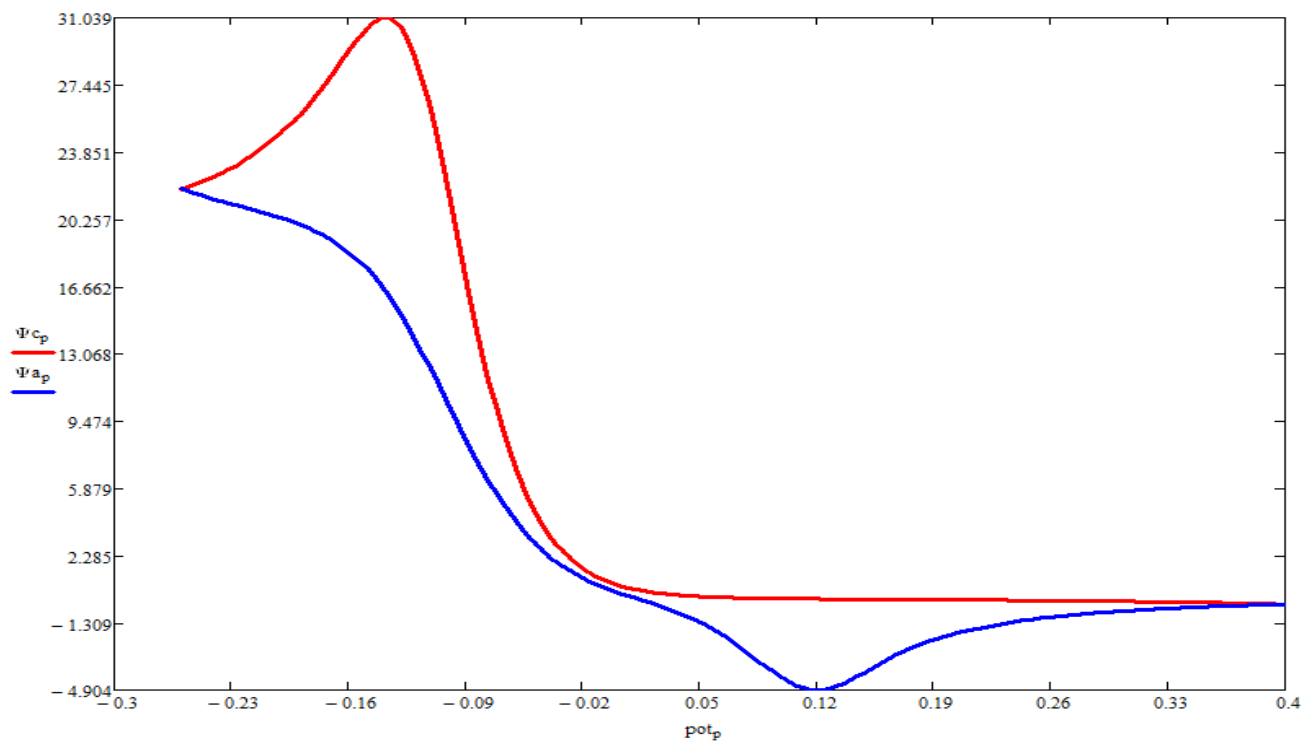
$$\Psi_n := \frac{\text{KET} \cdot e^{-\alpha \cdot \Phi_n} \left[ 1 - \frac{(1 + e^{\Phi_n}) \cdot \text{Keq}}{\text{Kcatalytic} \cdot (1 + \text{Keq}) \cdot 25} \cdot \sum_{j=1}^{n-1} (\Psi_j \cdot S_{n-j+1}) - \frac{(1 + e^{\Phi_n}) \cdot 1}{25 \cdot (1 + \text{Keq})} \cdot \sum_{j=1}^{n-1} (\Psi_j \cdot S_{n-j+1}) \right]}{1 + \text{KET} \cdot e^{-\alpha \cdot \Phi_n} \cdot (1 + e^{\Phi_n}) \cdot \frac{S_1 \cdot \text{Keq}}{\text{Kcatalytic} \cdot (1 + \text{Keq}) \cdot 25} + \text{KET} \cdot e^{-\alpha \cdot \Phi_n} \cdot (1 + e^{\Phi_n}) \cdot \frac{S_1 \cdot 1}{25 \cdot (1 + \text{Keq})}}$$

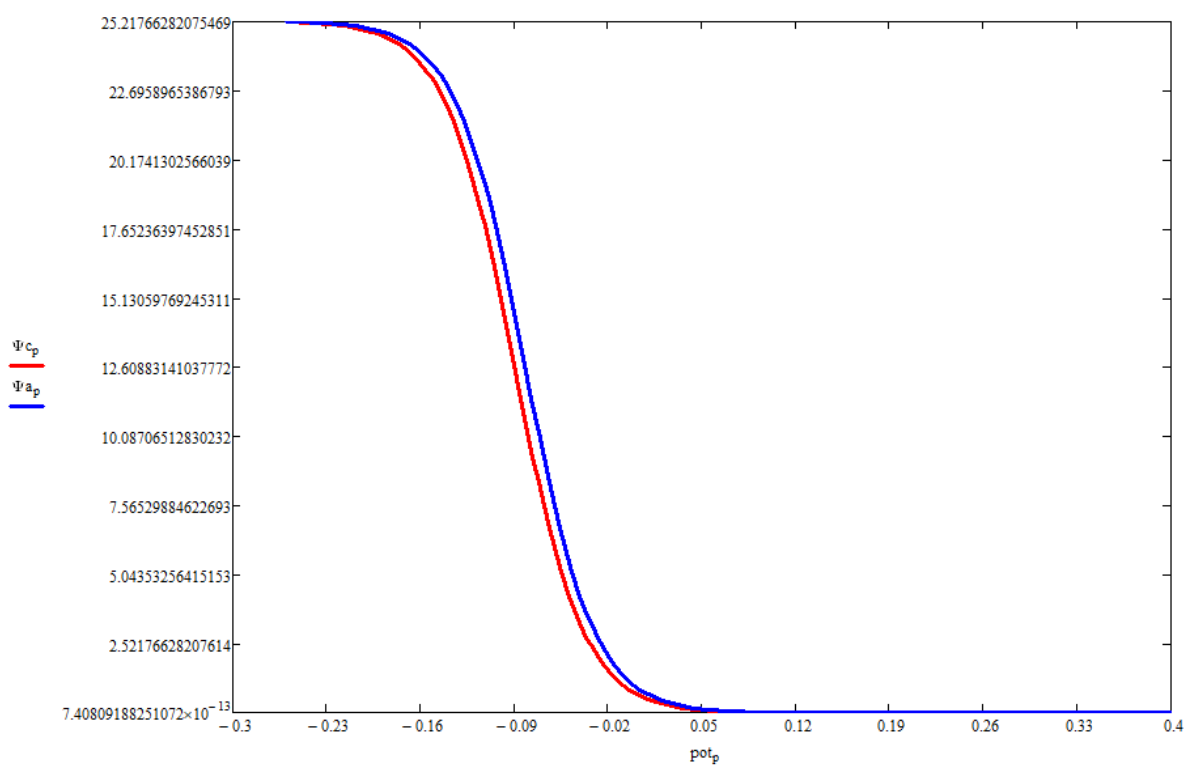
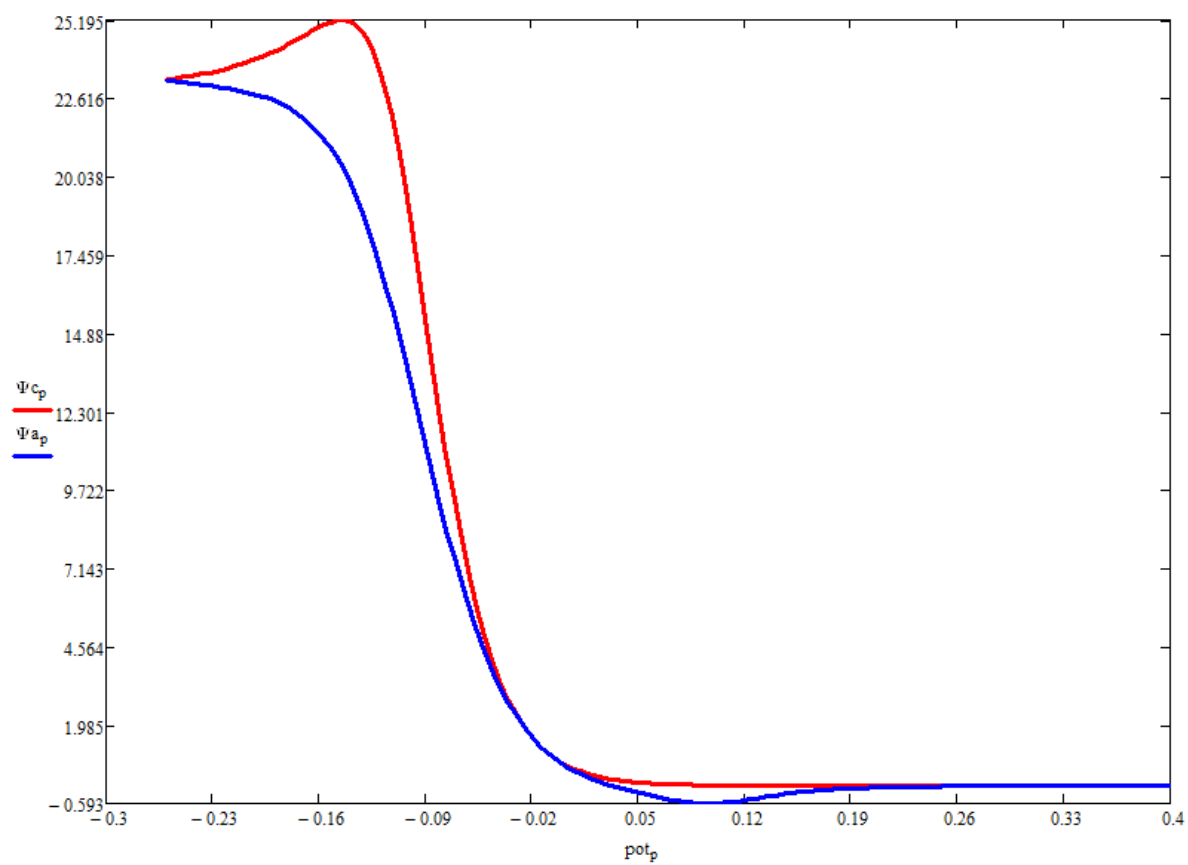
$$p := 35 \cdot \frac{\Delta E}{dE}$$

$$\Psi_{a_p} := (\Psi) \left( \frac{\tau}{d \cdot 25} + p \right) \cdot 25$$

$$\text{pot}_p := E_s + p \cdot dE$$

$$\Psi_{c_p} := (\Psi) \left[ \left[ \frac{\Delta E}{dE} \cdot 2 + \left( \frac{\tau}{25 \cdot d} \right) \right] - p \right] \cdot 25$$





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